



# Approximation of bicarbonate concentration using serum total carbon dioxide concentration in patients with non-dialysis chronic kidney disease

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**Background:** We investigated the relationship between serum total carbon dioxide (CO<sub>2</sub>) and bicarbonate ion (HCO<sub>3</sub><sup>-</sup>) concentrations in pre-dialysis chronic kidney disease (CKD) patients and devised a formula for predicting low bicarbonate (HCO<sub>3</sub><sup>-</sup> < 24 mmol/L) and high bicarbonate (HCO<sub>3</sub><sup>-</sup> ≥ 24 mmol/L) using clinical parameters.

**Methods:** In total, 305 samples of venous blood collected from 207 pre-dialysis patients assessed by CKD stage (G1 + G2, 46; G3, 50; G4, 51; G5, 60) were investigated. The relationship between serum total CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> concentrations was analyzed using Pearson's correlation coefficient. An approximation formula was developed using clinical parameters correlated independently with HCO<sub>3</sub><sup>-</sup> concentration. Diagnostic accuracy of serum total CO<sub>2</sub> and the approximation formula was evaluated by receiver operating characteristic curve analysis and a 2 × 2 table.

**Results:** Serum total CO<sub>2</sub> correlated strongly with HCO<sub>3</sub><sup>-</sup> concentration ( $r = 0.91$ ;  $P < 0.001$ ). The following approximation formula was obtained by a multiple linear regression analysis: HCO<sub>3</sub><sup>-</sup> (mmol/L) = total CO<sub>2</sub> – 0.5 × albumin – 0.1 × chloride – 0.01 × (estimated glomerular filtration rate + blood glucose) + 15. The areas under the curves of serum total CO<sub>2</sub> and the approximation formula for detection of low bicarbonate and high bicarbonate were 0.981, 0.996, 0.993, and 1.000, respectively. This formula had superior diagnostic accuracy compared with that of serum total CO<sub>2</sub> (86.6% vs. 81.3%).

**Conclusion:** Serum total CO<sub>2</sub> correlated strongly with HCO<sub>3</sub><sup>-</sup> concentration in pre-dialysis CKD patients. An approximation formula including serum total CO<sub>2</sub> showed superior diagnostic accuracy for low and high bicarbonate compared with serum total CO<sub>2</sub>.

**Keywords:** Acid base balance, Bicarbonates, Carbon dioxide, Chronic kidney disease

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## Introduction

Metabolic acidosis is a common complication of chronic kidney disease (CKD) and can lead to bone mineral loss, protein wasting, progression of renal dysfunction, and higher mortality risk [1–3]. Therefore, early detection and accurate diagnosis of metabolic acidosis are important to prevent CKD progression and increased risk of mortality.

In Japan, blood-gas analyzers are available in most hospitals. Therefore, clinical practice guidelines for CKD recommend measurement of bicarbonate ions ( $\text{HCO}_3^-$ ) using samples of arterial/venous blood gases for assessment of metabolic acidosis in pre-dialysis CKD patients [4]. However, for these blood-gas analyses, a specific measurement device and syringe are required in addition to the blood samples used for biochemical analyses [5].

Serum total carbon-dioxide concentration (serum total  $\text{CO}_2$ ) can be measured readily, along with creatinine, urea, and electrolytes, using a biochemical analyzer in clinical settings [6]. Furthermore, values based on this measurement have been shown to be correlated strongly with  $\text{HCO}_3^-$  concentration in patients without renal impairment [7]. However, few studies have examined the relationship between serum total  $\text{CO}_2$  and  $\text{HCO}_3^-$  concentration in patients with renal impairment. Therefore, we analyzed the relationship between these two parameters in CKD patients who were not undergoing renal replacement therapy. Furthermore, we developed a new formula for approximation of  $\text{HCO}_3^-$  concentration using clinical parameters that included serum total  $\text{CO}_2$  and evaluated the diagnostic accuracy of the approximated values derived from this new formula.

## Methods

### *Ethical approval of the study protocol*

This study was carried out in accordance with the ethical principles contained within the Declaration of Helsinki. The study protocol was approved by the Ethics Committee of Saitama Medical Center, Jichi Medical University (No. S17-052; Saitama, Japan). The requirement of informed consent was waived, and an opt-out method was used due to the retrospective design of the study.

### *Inclusion and exclusion criteria*

Inclusion criteria were (i) age > 20 years; (ii) stable outpatient with CKD stage G1 to 5; and (iii) simultaneous measurement of serum total  $\text{CO}_2$  and  $\text{HCO}_3^-$  concentration. Exclusion criteria were patients (i) undergoing dialysis therapy and (ii) who had undergone renal transplantation.

### *Study design*

This was a single-center, retrospective, cross-sectional study. We analyzed patient data obtained from medical records from the Division of Nephrology, Saitama Medical Center, between April 2016 and March 2018. Laboratory data of blood tests and blood-gas tests obtained simultaneously were used for analyses.

The relationship between serum total  $\text{CO}_2$  and  $\text{HCO}_3^-$  concentration was analyzed using Pearson's correlation coefficient. An approximation formula was developed by multiple linear regression analysis with independent factors correlated with  $\text{HCO}_3^-$  concentration. The relationship between  $\text{HCO}_3^-$  concentration approximated by our formula and actual  $\text{HCO}_3^-$  concentration was analyzed using Pearson's correlation coefficient. The diagnostic accuracy of serum total  $\text{CO}_2$  and approximated  $\text{HCO}_3^-$  concentration for low and high bicarbonate levels was analyzed using receiver operating characteristic (ROC) curve analysis and a  $2 \times 2$  table.

### *Laboratory methods*

Blood and urinary parameters were determined by the Department of Clinical Laboratory, Saitama Medical Center. Samples of venous blood were collected in ethylenediamine tetraacetic acid (EDTA)-containing tubes from the brachial vein and centrifuged within 15 minutes to obtain serum. Serum total  $\text{CO}_2$  was measured within 15 minutes after centrifugation using an automated biochemical analyzer (JCA-BM6070; JEOL, Tokyo, Japan), as were the biochemical parameters hemoglobin, total protein, serum albumin, blood urea nitrogen, serum creatinine, sodium, potassium, chloride, calcium, phosphate, magnesium, and glucose. Serum total  $\text{CO}_2$  was determined by an enzymatic method using a commercial kit (Toyobo, Osaka, Japan).

Samples of venous blood for gas analyses were collected in a heparinized blood-gas syringe from the brachial vein simultaneously with samples for other blood tests. These samples were analyzed within 10 minutes to obtain values for pH and partial pressure of carbon dioxide (pCO<sub>2</sub>). Blood pH and pCO<sub>2</sub> were measured using a blood-gas analyzer (Rapidlab-1265; Siemens Healthcare Diagnostics, Tarrytown, NY, USA). The HCO<sub>3</sub><sup>-</sup> concentration was calculated from measured pH and pCO<sub>2</sub> using the Henderson–Hasselbalch equation [8]:

$$\text{pH} = 6.1 + \log([\text{HCO}_3^-]/\text{pCO}_2 \times 0.03).$$

The estimated glomerular filtration rate (eGFR) was calculated using a modified version of the Modification of Diet in Renal Disease formula set by the Japanese Society of Nephrology [9]:

$$\text{eGFR (mL/min/1.73 m}^2) = 194 \times \text{age}^{-0.287} \times \text{serum creatinine}^{-1.094} \text{ (multiplied by 0.739 for females).}$$

### Statistical analyses

Statistical analyses were performed using JMP v11 (SAS Institute, Cary, NC, USA). Data are the mean ± standard deviation for continuous variables and are count and percentage for categorical variables. Comparisons of component ratios among groups were performed using Fisher’s exact test with the Bonferroni correction. Comparisons of clinical parameters among groups were

performed using the Kruskal–Wallis test with the Steel–Dwass test. Correlations between two variables were evaluated by Pearson’s correlation coefficient. Linear regression analysis was used to detect factors independently correlated with HCO<sub>3</sub><sup>-</sup> concentration. Parameters that significantly correlated with HCO<sub>3</sub><sup>-</sup> concentration in a simple linear regression analysis were included in a multiple linear regression analysis. An approximation formula involving serum total CO<sub>2</sub> was determined using variables that independently correlated with HCO<sub>3</sub><sup>-</sup> concentration in the multiple linear regression analysis. The diagnostic accuracy of serum total CO<sub>2</sub> and approximated HCO<sub>3</sub><sup>-</sup> concentration was examined using ROC curve analysis and a 2 × 2 table. The area under the curve (AUC), sensitivity, specificity, positive predictive value, negative predictive value, and accuracy were calculated for detection of low bicarbonate (HCO<sub>3</sub><sup>-</sup> < 24 mmol/L) and high bicarbonate (HCO<sub>3</sub><sup>-</sup> ≥ 24 mmol/L). For all tests, *P* < 0.05 was considered significant.

## Results

### Patient characteristics

The characteristics of patients and their medications were categorized by CKD stage (G1 + G2, G3, G4, G5) (Table 1). A total of 305 blood samples from 207 patients

**Table 1.** Comparison of patient characteristics and medications according to chronic kidney disease (CKD) stage

CKD stage	All	G1 + G2	G3	G4	G5	<i>P</i> value
Number of patients	207	46	50	51	60	
Number of samples	305	59	65	70	111	
Age (yr)	64.7 ± 16.1	49.2 ± 18.1*	68.0 ± 11.5	71.1 ± 12.1	68.4 ± 13.4	* <i>P</i> < 0.001 vs. G3, G4 and G5
Male	139 (67.1)	24 (52.2)	35 (70.0)	37 (72.5)	43 (71.7)	
Body mass index (kg/m <sup>2</sup> )	23.8 ± 5.1	23.4 ± 4.8	24.1 ± 5.1	23.6 ± 3.8	24.0 ± 6.3	
Diabetes mellitus	75 (36.2)	9 (19.6)	20 (40.0)	22 (43.1)	24 (40.0)	
Corticosteroid	30 (14.5)	13 (28.3)	9 (18.0)	5 (9.8)	3 (5.0)*	* <i>P</i> < 0.05 vs. G1 + G2
β-blocker	46 (22.2)	1 (2.2)	7 (14.0)	15 (29.4)*	23 (38.3)**	* <i>P</i> < 0.005 vs. G1 + G2, ** <i>P</i> < 0.001 vs. G1 + G2, <i>P</i> < 0.05 vs. G3
Renin–angiotensin system inhibitor	131 (63.3)	19 (41.3)*	31 (62.0)	38 (74.5)	43 (71.7)	* <i>P</i> < 0.01 vs. G4, <i>P</i> < 0.05 vs. G5
Aldosterone receptor antagonist	7 (3.4)	1 (2.2)	2 (4.0)	3 (5.9)	1 (1.7)	
Loop diuretics	64 (30.9)	0 (0.0)*	14 (28.0)	19 (37.3)	31 (51.7)	* <i>P</i> < 0.001 vs. G3, G4, and G5
Thiazide diuretics	16 (7.7)	2 (4.3)	0 (0.0)	2 (3.9)	12 (20.0)*	* <i>P</i> < 0.005 vs. G3
Sodium bicarbonate	37 (17.9)	2 (4.3)	6 (12.0)	7 (13.7)	22 (36.7)*	* <i>P</i> < 0.001 vs. G1 + G2, <i>P</i> < 0.05 vs. G3
Potassium binder	23 (11.1)	1 (2.2)	3 (6.0)	5 (9.8)	14 (23.3)*	* <i>P</i> < 0.01 vs. G1 + G2
Phosphate binder	10 (4.8)	0 (0.0)	0 (0.0)	1 (2.0)	9 (15.0)*	* <i>P</i> < 0.05 vs. G1 + G2 and G3

Data are presented as number only, mean ± standard deviation, or number (%).

(139 males, 68 females; mean age,  $64.7 \pm 16.1$  years) was obtained. The number of patients for each CKD stage was 46 for G1 + G2, 50 for G3, 51 for G4, and 60 for G5.

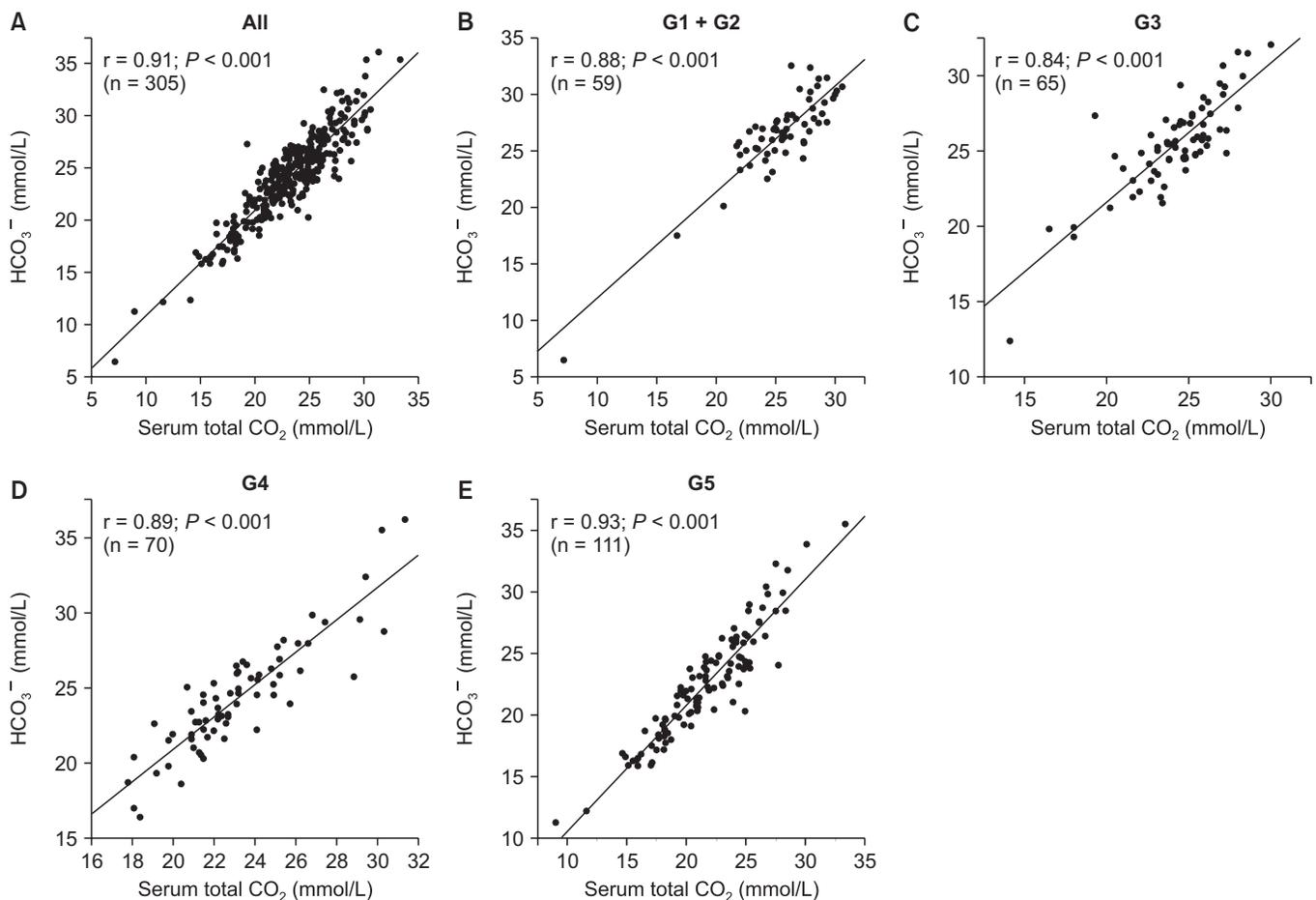
There were significant differences in age as well as use of corticosteroid,  $\beta$ -blocker, loop diuretic, sodium bicarbonate, potassium binder, phosphate binder, or inhibitor of the renin–angiotensin system (RAS) among groups categorized by CKD stage. Age was significantly lower in patients with CKD stage G1 + G2 than those with CKD stage G3, G4, or G5 ( $P < 0.001$  for all).

The proportions of patients using a  $\beta$ -blocker ( $P < 0.005$ , G1 + G2 vs. G4 and G5), RAS inhibitor ( $P < 0.05$ , G1 + G2 vs. G4 and G5), loop diuretic ( $P < 0.001$ , G1 + G2 vs. G3, G4, and G5), thiazide diuretic ( $P < 0.005$ , G3 vs. G5), sodium bicarbonate ( $P < 0.05$ , G5 vs. G1 + G2 and G3), potassium binder ( $P < 0.01$ , G1 + G2 vs. G5), or phosphate binder ( $P < 0.05$ , G5 vs. G1 + G2 and G3) increased

significantly with CKD stage. The percentage of patients using a corticosteroid decreased with CKD stage ( $P < 0.05$ , G1 + G2 vs. G5).

#### Correlation between serum total $\text{CO}_2$ and $\text{HCO}_3^-$ concentration

Fig. 1 shows the correlation between serum total  $\text{CO}_2$  and  $\text{HCO}_3^-$  concentration in each group categorized by CKD stage. Serum total  $\text{CO}_2$  was correlated with  $\text{HCO}_3^-$  concentration significantly and substantially in all patients ( $r = 0.91$ ;  $P < 0.001$ ), patients with CKD stage G1 + G2 ( $r = 0.88$ ;  $P < 0.001$ ), patients with CKD stage G3 ( $r = 0.84$ ;  $P < 0.001$ ), patients with CKD stage G4 ( $r = 0.89$ ;  $P < 0.001$ ), and patients with CKD stage G5 ( $r = 0.93$ ;  $P < 0.001$ ).



**Figure 1.** Relationships between serum total carbon dioxide ( $\text{CO}_2$ ) and measured bicarbonate ion ( $\text{HCO}_3^-$ ) concentration according to chronic kidney disease (CKD) stage. (A) All patients, (B) patients with CKD stage G1 + G2, (C) patients with CKD stage G3, (D) patients with CKD stage G4, (E) patients with CKD stage G5.

*Formula for approximation of HCO<sub>3</sub><sup>-</sup> concentration*

A simple linear regression analysis showed that HCO<sub>3</sub><sup>-</sup> concentration was significantly negatively correlated with age, blood urea nitrogen, potassium, chloride, phosphate, blood glucose, and use of an RAS inhibitor, sodium bicarbonate, potassium binder, or phosphate binder. The HCO<sub>3</sub><sup>-</sup> concentration showed significant positive correlations with serum albumin, hemoglobin, eGFR, total calcium, serum total CO<sub>2</sub>, and use of a corticosteroid or aldosterone receptor antagonist. We performed a multivariate linear regression analysis using variables that showed a significant correlation with HCO<sub>3</sub><sup>-</sup> concentration in the

simple linear regression analysis (Table 2). The multiple linear regression analysis revealed that serum albumin ( $P = 0.006$ ), eGFR ( $P = 0.047$ ), chloride ( $P < 0.001$ ), blood glucose ( $P = 0.004$ ), and serum total CO<sub>2</sub> ( $P < 0.001$ ) were independently correlated with HCO<sub>3</sub><sup>-</sup> concentration.

An approximated HCO<sub>3</sub><sup>-</sup> formula based on serum total CO<sub>2</sub> was developed using variables that showed a significant correlation with HCO<sub>3</sub><sup>-</sup> concentration in the multiple linear regression analysis:

Approximated HCO<sub>3</sub><sup>-</sup> concentration (mmol/L) = (0.859 × total CO<sub>2</sub>) – (0.559 × albumin) – (0.145 × chloride) – (0.00920 × eGFR) – (0.00721 × blood glucose) + 23.070.

Inputting the mean values of serum albumin, eGFR,

**Table 2.** Simple and multiple linear regression analyses of the variables correlated with HCO<sub>3</sub><sup>-</sup> concentration

Variable	Simple linear regression analysis		Multivariate linear regression analysis		
	Coefficient (β)	P value	Coefficient (β)	Standard coefficient	P value
Constant			23.070		
Age (yr)	-0.036	0.017	-0.005	-0.019	0.50
Male (yes vs. no)	-0.913	0.08			
Body mass index (kg/m <sup>2</sup> )	-0.060	0.18			
Diabetes mellitus (yes vs. no)	0.105	0.83			
Corticosteroid (yes vs. no)	2.261	< 0.001	-0.011	-0.001	0.97
β-blocker (yes vs. no)	-0.878	0.11			
Renin-angiotensin system inhibitor (yes vs. no)	-1.568	0.002	0.015	0.002	0.94
Aldosterone receptor antagonist (yes vs. no)	2.857	0.033	-0.185	-0.008	0.73
Loop diuretic (yes vs. no)	0.081	0.87			
Thiazide diuretic (yes vs. no)	0.450	0.54			
Sodium bicarbonate (yes vs. no)	-3.287	< 0.001	-0.455	-0.043	0.08
Potassium binder (yes vs. no)	-1.889	0.007	0.022	0.002	0.94
Phosphate binder (yes vs. no)	-3.472	< 0.001	0.298	0.019	0.45
Total protein (g/dL)	1.009	0.003			
Serum albumin (g/dL)	1.464	< 0.001	-0.559	-0.077	0.006
Hemoglobin (g/dL)	0.800	< 0.001	0.078	0.042	0.19
Blood urea nitrogen (mg/dL)	-0.050	< 0.001	-0.012	-0.080	0.06
Creatinine (mg/dL)	-0.702	< 0.001			
Estimated glomerular filtration rate (mL/min/1.73 m <sup>2</sup> )	0.047	< 0.001	-0.009	-0.069	0.047
Uric acid (mg/dL)	-0.052	0.28			
Sodium (mmol/L)	-0.011	0.89			
Potassium (mmol/L)	-3.587	< 0.001	-0.227	-0.029	0.28
Chloride (mmol/L)	-0.480	< 0.001	-0.145	-0.172	< 0.001
Total calcium (mg/dL)	2.360	< 0.001	0.133	0.024	0.42
Phosphate (mg/dL)	-1.240	< 0.001	-0.064	-0.020	0.59
Magnesium (mg/dL)	0.154	0.83			
Blood glucose (mg/dL)	-0.011	0.047	-0.007	-0.072	0.004
Serum total CO <sub>2</sub> (mmol/L)	1.009	< 0.001	0.859	0.777	< 0.001

CO<sub>2</sub>, carbon dioxide; HCO<sub>3</sub><sup>-</sup>, bicarbonate ion.

chloride, blood glucose, and serum total  $\text{CO}_2$  into the model, the formula was simplified into a final version [10]:

Approximated  $\text{HCO}_3^-$  concentration (mmol/L) = total  $\text{CO}_2$  – (0.5 × albumin) – (0.1 × chloride) – [0.01 × (eGFR + blood glucose)] + 15.

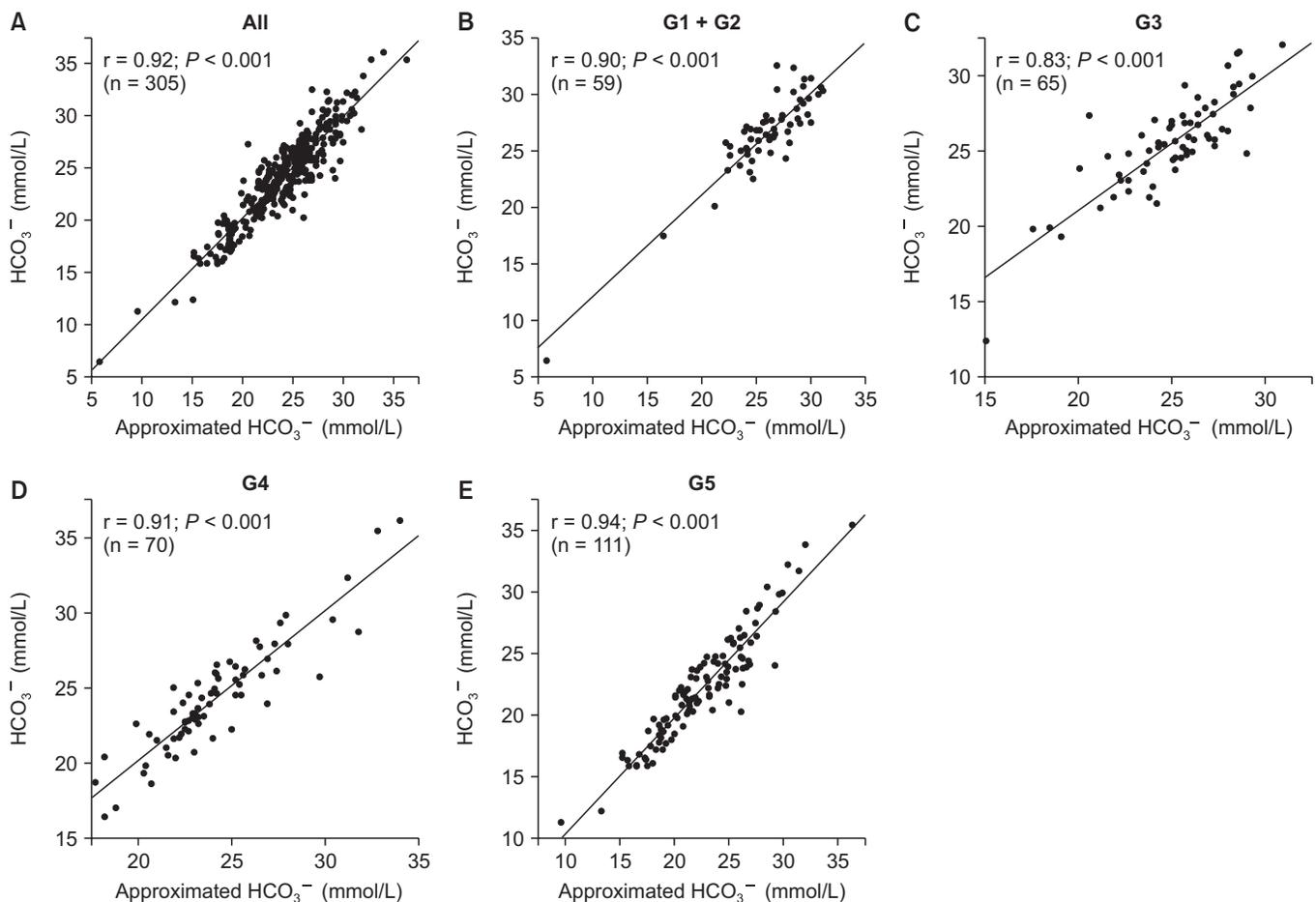
#### Correlation between approximated $\text{HCO}_3^-$ and measured $\text{HCO}_3^-$ concentrations

Fig. 2 shows the correlation between approximated  $\text{HCO}_3^-$  concentration calculated by our formula and measured  $\text{HCO}_3^-$  concentration in each group categorized by CKD stage. The approximated  $\text{HCO}_3^-$  concentration was significantly correlated with the measured  $\text{HCO}_3^-$  concentration in all patients ( $r = 0.92$ ;  $P < 0.001$ ), patients with CKD stage G1 + G2 ( $r = 0.90$ ;  $P < 0.001$ ), patients with

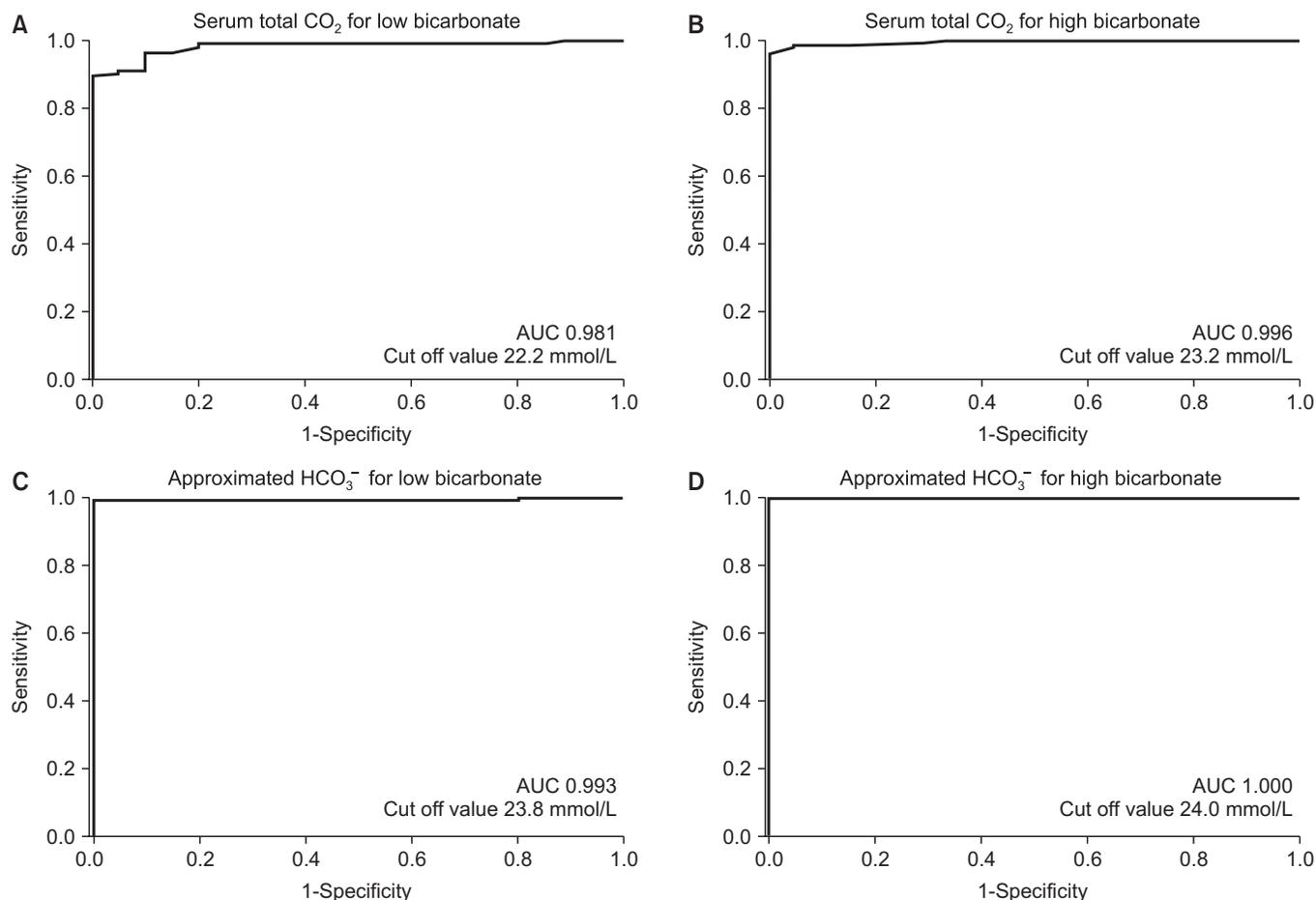
CKD stage G3 ( $r = 0.83$ ;  $P < 0.001$ ), patients with CKD stage G4 ( $r = 0.91$ ;  $P < 0.001$ ), and patients with CKD stage G5 ( $r = 0.94$ ;  $P < 0.001$ ).

#### Diagnostic accuracy of serum total $\text{CO}_2$ and approximated $\text{HCO}_3^-$ concentration for prediction of low bicarbonate and high bicarbonate levels

The ROC curves of serum total  $\text{CO}_2$  and approximated  $\text{HCO}_3^-$  concentration for detecting low bicarbonate ( $\text{HCO}_3^- < 24$  mmol/L) and high bicarbonate ( $\text{HCO}_3^- \geq 24$  mmol/L) are shown in Fig. 3. The AUCs of serum total  $\text{CO}_2$  and approximated  $\text{HCO}_3^-$  concentration for detection of low bicarbonate and high bicarbonate were 0.981, 0.996, 0.993, and 1.000, respectively. The optimal cut-off values of serum total  $\text{CO}_2$  and approximated  $\text{HCO}_3^-$  concentration for detection of low bicarbonate and



**Figure 2.** Relationships between approximated bicarbonate ion ( $\text{HCO}_3^-$ ) concentration and measured  $\text{HCO}_3^-$  concentration according to chronic kidney disease (CKD) stage. (A) All patients, (B) patients with CKD stage G1 + G2, (C) patients with CKD stage G3, (D) patients with CKD stage G4, (E) patients with CKD stage G5.



**Figure 3.** Receiver operating characteristic (ROC) curve analysis for detecting low bicarbonate (bicarbonate ion  $[HCO_3^-] < 24$  mmol/L) and high bicarbonate ( $HCO_3^- \geq 24$  mmol/L). (A) The ROC curve of serum total carbon dioxide ( $CO_2$ ) for low bicarbonate. (B) The ROC curve of serum total  $CO_2$  for high bicarbonate. (C) The ROC curve of approximated  $HCO_3^-$  concentration for low bicarbonate. (D) The ROC curve of approximated  $HCO_3^-$  concentration for high bicarbonate. AUC, area under the curve.

high bicarbonate were 22.2 mmol/L, 23.2 mmol/L, 23.8 mmol/L, and 24.0 mmol/L, respectively. The  $2 \times 2$  tables stratified by serum total  $CO_2$ , the approximated  $HCO_3^-$  concentration, and the measured  $HCO_3^-$  concentration for low bicarbonate and high bicarbonate are shown in Table 3. The diagnostic accuracy values of serum total  $CO_2$  and approximated  $HCO_3^-$  concentration for prediction of low bicarbonate and high bicarbonate are shown in Table 4. The sensitivity, specificity, positive predictive value, negative predictive value, accuracy, pre-test probability, positive post-test probability, and negative post-test probability were 91.7%, 73.4%, 72.5%, 92.0%, 81.3%, 43.3%, 72.5%, and 8.0% for serum total  $CO_2$  and 84.8%, 87.9%, 84.2%, 88.4%, 86.6%, 43.3%, 84.2%, and 11.6% for approximated  $HCO_3^-$  concentration, respectively. The

approximated  $HCO_3^-$  concentration showed superior accuracy compared with that for serum total  $CO_2$  (86.6% vs. 81.3%).

### Discussion

We assessed the relationship between serum total  $CO_2$  and  $HCO_3^-$  concentration in CKD patients who were not undergoing renal replacement therapy. This assessment enabled development of an approximation formula for prediction of low bicarbonate and high bicarbonate using clinical parameters involving serum total  $CO_2$ .

Serum total  $CO_2$  is the total concentration of all forms of  $CO_2$  in a serum sample:  $HCO_3^-$ , carbonate, and dissolved  $CO_2$ . In general, serum total  $CO_2$  value is approximately

**Table 3.** 2 × 2 tables stratified by serum total CO<sub>2</sub>, approximated HCO<sub>3</sub><sup>-</sup> concentration and measured HCO<sub>3</sub><sup>-</sup> concentration for low bicarbonate (HCO<sub>3</sub><sup>-</sup> < 24 mmol/L) and high bicarbonate (HCO<sub>3</sub><sup>-</sup> ≥ 24 mmol/L)

	HCO <sub>3</sub> <sup>-</sup>		Total
	Low bicarbonate (HCO <sub>3</sub> <sup>-</sup> < 24 mmol/L)	High bicarbonate (HCO <sub>3</sub> <sup>-</sup> ≥ 24 mmol/L)	
Serum total CO <sub>2</sub>			
Low serum total CO <sub>2</sub> (serum total CO <sub>2</sub> < 24 mmol/L)	121	46	167
High serum total CO <sub>2</sub> (serum total CO <sub>2</sub> ≥ 24 mmol/L)	11	127	138
Total	132	173	305
Approximated HCO <sub>3</sub> <sup>-</sup>			
Low approximated HCO <sub>3</sub> <sup>-</sup> (approximated HCO <sub>3</sub> <sup>-</sup> < 24 mmol/L)	112	21	133
High approximated HCO <sub>3</sub> <sup>-</sup> (approximated HCO <sub>3</sub> <sup>-</sup> ≥ 24 mmol/L)	20	152	172
Total	132	173	305

CO<sub>2</sub>, carbon dioxide; HCO<sub>3</sub><sup>-</sup>, bicarbonate ion.

**Table 4.** Comparison of diagnostic values between serum total CO<sub>2</sub> and approximated HCO<sub>3</sub><sup>-</sup>

	Serum total CO <sub>2</sub>	Approximated HCO <sub>3</sub> <sup>-</sup>
Sensitivity (%)	91.7	84.8
Specificity (%)	73.4	87.9
Positive predictive value (%)	72.5	84.2
Negative predictive value (%)	92.0	88.4
Accuracy (%)	81.3	86.6
Pre-test probability (%)	43.3	43.3
Positive post-test probability (%)	72.5	84.2
Negative post-test probability (%)	8.0	11.6

CO<sub>2</sub>, carbon dioxide; HCO<sub>3</sub><sup>-</sup>, bicarbonate ion.

equivalent to the HCO<sub>3</sub><sup>-</sup> concentration because most of the CO<sub>2</sub> in blood exists as HCO<sub>3</sub><sup>-</sup> [6]. Furthermore, serum total CO<sub>2</sub> was shown to have a substantial correlation with HCO<sub>3</sub><sup>-</sup> concentration by Kumar and Karon [7]. However, a discrepancy between serum total CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> concentration caused by the influence of temperature and acidity [11] is occasionally observed in patients without renal impairment [12]. In the present study, serum albumin, eGFR, chloride, and blood glucose, in addition to serum total CO<sub>2</sub>, were independently associated with HCO<sub>3</sub><sup>-</sup> concentration in serum.

Increased serum albumin has been reported to be associated with metabolic acidosis in pre-dialysis CKD patients [13], and this phenomenon can be explained, at least in part, by the weak acidity of albumin [14]. These findings are consistent with our result showing a negative correlation between the concentrations of albumin and HCO<sub>3</sub><sup>-</sup> in serum. Furthermore, the HCO<sub>3</sub><sup>-</sup> concentra-

tion in serum decreases with progression of CKD stage [13], and this reduction has been suggested to be due to the inability of the kidney to synthesize ammonia, regenerate HCO<sub>3</sub><sup>-</sup>, and excrete hydrogen ions (H<sup>+</sup>) [15]. Therefore, HCO<sub>3</sub><sup>-</sup> concentration was expected to have a positive correlation with eGFR. However, in the present study, HCO<sub>3</sub><sup>-</sup> concentration was negatively correlated with GFR. This difference between our result and those of published reports may be explained by the increase in the ratio of use of diuretics and sodium bicarbonate with progression of CKD stage because such use leads to an increase in HCO<sub>3</sub><sup>-</sup> concentration in serum [13,16]. Loop diuretics and thiazide diuretics inhibit the Na<sup>+</sup>-K<sup>+</sup>-2Cl<sup>-</sup> cotransporter and Na<sup>+</sup>-Cl<sup>-</sup> cotransporter, respectively, and increase sodium delivery to distal tubular segments. The delivered sodium is reabsorbed at cortical collecting ducts due to, at least in part, the increase of serum aldosterone induced by diuretic-associated reduction of intravascular fluids [17]. Furthermore, an increase in serum aldosterone associated with diuretic use stimulates H<sup>+</sup>-ATPase activity at cortical collecting ducts [18], which leads to an increase in HCO<sub>3</sub><sup>-</sup> concentration in serum. An increase in serum aldosterone also stimulates potassium excretion at cortical collecting ducts, which leads to a decrease in serum potassium concentration [19]. A decreased serum potassium concentration may increase renal production of ammonia and excretion of ammonium ions, which result in an increase in HCO<sub>3</sub><sup>-</sup> concentration in serum [20]. In addition, sodium bicarbonate is administered frequently in CKD patients with metabolic acidosis. Therefore, use of diuretics and sodium bicarbonate may reflect the inverse relationship between changes in

eGFR and  $\text{HCO}_3^-$  concentration in serum noted in our study. However, further studies in a much larger cohort are needed to confirm the relationship between these factors.

Hyperchloremic metabolic acidosis is observed in 30% to 50% of patients with chronic renal failure [21]. The chloride concentration in serum has been reported to increase as  $\text{HCO}_3^-$  concentration in serum decreases [22]. We documented a negative correlation between chloride and  $\text{HCO}_3^-$  concentrations in serum with progression of CKD stage, a finding that is compatible with that of Widmer et al [22]. Blood glucose was negatively correlated with  $\text{HCO}_3^-$  concentration in serum in the present study. Uremia inhibits insulin secretion as well as insulin sensitivity [23], which leads to an increase in blood glucose concentration with metabolic acidosis via reduction in  $\text{Na}^+/\text{H}^+$  exchanger activity [24]. Glucose appears in urine if the blood glucose concentration exceeds the renal threshold of 170 to 200 mg/dL [25]. Increased urinary glucose has been shown to inhibit  $\text{H}^+$  excretion through proximal renal tubules [26] and activates the sodium–glucose-coupled transporter, which inhibits the  $\text{Na}^+/\text{H}^+$  exchanger via competition for sodium influx. The subsequent decreased  $\text{H}^+$  excretion leads to metabolic acidosis with a reduction of  $\text{HCO}_3^-$  concentration in plasma [27]. Therefore, our findings may be explained by data reported previously.

Measurement of serum total  $\text{CO}_2$  has two main advantages compared with blood-gas analyses. First, the cost of a blood-gas syringe can be saved, and the amount of blood collected is reduced. Second, serum total  $\text{CO}_2$  can be used to predict metabolic acidosis and metabolic alkalosis without use of a blood-gas analyzer. Therefore, measurement of serum total  $\text{CO}_2$  would alleviate some of the burden on patients and laboratory staff. In addition, the approximated  $\text{HCO}_3^-$  concentration derived from clinical parameters, including serum total  $\text{CO}_2$ , could have been useful for predicting disturbances of acid–base metabolism in the present study.

Our study had four main limitations. First, this was a single-center, retrospective, observational study and may have been subject to bias in patient selection. Second, the study cohort was small, which limits the generalizability of our findings. Third, several baseline characteristics, including age and medication use, were significantly different among groups categorized by CKD stage.

Fourth, we used venous blood samples for analyses. The results might have been different if samples of arterial blood had been used. However, pH and  $\text{HCO}_3^-$  have been reported to show sufficient agreement between arterial and venous blood-gas analysis [28]. Therefore, further prospective, large-scale, multicenter studies with arterial blood samples for gas analysis are required to confirm our findings.

In conclusion, serum total  $\text{CO}_2$  was substantially correlated with  $\text{HCO}_3^-$  concentration in the serum of pre-dialysis CKD patients. An approximation formula including serum total  $\text{CO}_2$  showed superior diagnostic accuracy for low and high bicarbonate levels compared with serum total  $\text{CO}_2$ .

### Conflicts of interest

All authors have no conflicts of interest to declare.

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### Authors' contributions

Keiji Hirai and Susumu Ookawara conceived and designed the research. Keiji Hirai, Haruhisa Miyazawa, Kiyonori Ito, Yuichirou Ueda, Yoshio Kaku, and Taro Hoshino performed research. Saori Minato, Shohei Kaneko, Katsunori Yanai, Hiroki Ishii, Taisuke Kitano, and Mitsutoshi Shindo collected the data. Keiji Hirai, Tatsuro Watano, Shinji Fujino, and Kiyoka Omoto performed the analysis. Keiji Hirai and Susumu Ookawara wrote the paper. Yoshiyuki Morishita made critical revisions and approved the final version. All authors read and approved the final manuscript.

### References

- [1] Cochran M, Wilkinson R. Effect of correction of metabolic acidosis on bone mineralisation rates in patients with renal osteomalacia. *Nephron* 1975;15:98-110.

- [2] Ballmer PE, McNurlan MA, Hulter HN, Anderson SE, Glick PJ, Krapf R. Chronic metabolic acidosis decreases albumin synthesis and induces negative nitrogen balance in humans. *J Clin Invest* 1995;95:39-45.
- [3] Raphael KL, Wei G, Baird BC, Greene T, Beddhu S. Higher serum bicarbonate levels within the normal range are associated with better survival and renal outcomes in African Americans. *Kidney Int* 2011;79:356-362.
- [4] Nihon Jinzo Gakkai. [Special issue: evidence-based practice guideline for the treatment of CKD]. *Nihon Jinzo Gakkai Shi* 2013;55:585-860. Japanese.
- [5] O'Leary TD, Langton SR. Calculated bicarbonate or total carbon dioxide? *Clin Chem* 1989;35:1697-1700.
- [6] Dobson GP, Veech RL, Hoeger U, Passonneau JV. Enzymatic determination of total CO<sub>2</sub> in freeze-clamped animal tissues and plasma. *Anal Biochem* 1991;195:232-237.
- [7] Kumar V, Karon BS. Comparison of measured and calculated bicarbonate values. *Clin Chem* 2008;54:1586-1587.
- [8] Ramsay AG. Clinical application of the Henderson-Hasselbalch equation. *Appl Ther* 1965;7:730-736.
- [9] Matsuo S, Imai E, Horio M, et al; Collaborators developing the Japanese equation for estimated GFR. Revised equations for estimated GFR from serum creatinine in Japan. *Am J Kidney Dis* 2009;53:982-992.
- [10] Jain A, Bhayana S, Vlasschaert M, House A. A formula to predict corrected calcium in haemodialysis patients. *Nephrol Dial Transplant* 2008;23:2884-2888.
- [11] Maas AH, van Heijst AN, Visser BF. The determination of the true equilibrium constant ( $pK_{ig}$ ) and the practical equilibrium coefficient ( $pK'_{ig}$ ) for the first ionization of carbonic acid in solutions of sodium bicarbonate, cerebrospinal fluid, plasma and serum at 25° and 38°. *Clin Chim Acta* 1971;33:325-343.
- [12] Kim Y, Massie L, Murata GH, Tzamaloukas AH. Discrepancy between measured serum total carbon dioxide content and bicarbonate concentration calculated from arterial blood gases. *Cureus* 2015;7:e398.
- [13] Raphael KL, Zhang Y, Ying J, Greene T. Prevalence of and risk factors for reduced serum bicarbonate in chronic kidney disease. *Nephrology (Carlton)* 2014;19:648-654.
- [14] Rossing TH, Maffeo N, Fencel V. Acid-base effects of altering plasma protein concentration in human blood in vitro. *J Appl Physiol* (1985) 1986;61:2260-2265.
- [15] Yaqoob MM. Acidosis and progression of chronic kidney disease. *Curr Opin Nephrol Hypertens* 2010;19:489-492.
- [16] de Brito-Ashurst I, Varaganam M, Raftery MJ, Yaqoob MM. Bicarbonate supplementation slows progression of CKD and improves nutritional status. *J Am Soc Nephrol* 2009;20:2075-2084.
- [17] Greenberg A. Diuretic complications. *Am J Med Sci* 2000;319:10-24.
- [18] Garg LC. Respective roles of H-ATPase and H-K-ATPase in ion transport in the kidney. *J Am Soc Nephrol* 1991;2:949-960.
- [19] Arroyo JP, Ronzaud C, Lagnaz D, Staub O, Gamba G. Aldosterone paradox: differential regulation of ion transport in distal nephron. *Physiology (Bethesda)* 2011;26:115-123.
- [20] Han KH. Mechanisms of the effects of acidosis and hypokalemia on renal ammonia metabolism. *Electrolyte Blood Press* 2011;9:45-49.
- [21] Enia G, Catalano C, Zoccali C, et al. Hyperchloraemia: a non-specific finding in chronic renal failure. *Nephron* 1985;41:189-192.
- [22] Widmer B, Gerhardt RE, Harrington JT, Cohen JJ. Serum electrolyte and acid base composition. The influence of graded degrees of chronic renal failure. *Arch Intern Med* 1979;139:1099-1102.
- [23] Alvestrand A, Mujagic M, Wajngot A, Efendic S. Glucose intolerance in uremic patients: the relative contributions of impaired beta-cell function and insulin resistance. *Clin Nephrol* 1989;31:175-183.
- [24] Lynch CJ, Wilson PB, Blackmore PF, Exton JH. The hormone-sensitive hepatic Na<sup>+</sup>-pump. Evidence for regulation by diacylglycerol and tumor promoters. *J Biol Chem* 1986;261:14551-14556.
- [25] Lawrence RD. Renal threshold for glucose: normal and in diabetics. *Br Med J* 1940;1:766-768.
- [26] Nascimento-Gomes G, Mello-Aires M. Effect of glucose on the kinetics of bicarbonate reabsorption in the early and middle proximal tubule. *Braz J Med Biol Res* 1990;23:79-85.
- [27] Lang F, Messner G, Rehwald W. Electrophysiology of sodium-coupled transport in proximal renal tubules. *Am J Physiol* 1986;250:F953-F962.
- [28] Kelly AM. Review article: can venous blood gas analysis replace arterial in emergency medical care. *Emerg Med Australas* 2010;22:493-498.